4 SEM TDC CHMH (CBCS) C 9

2023

(May/June)

CHEMISTRY

(Core)

Paper: C-9

(Organic Chemistry)

Full Marks: 53

Pass Marks: 21

Time: 3 hours

The figures in the margin indicate full marks for the questions

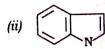
- 1. Choose the correct answer from the following:

 1×4=4
 - (a) Anthracene when reduced with sodium and ethanol gives
 - (i) 9,10-anthraquinone
 - (ii) 9,10-dihydroanthracene
 - (iii) anthrone
 - (iv) None of the above

P23/993

(b) Which of the following is not a heterocyclic compound?

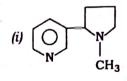


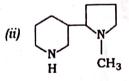


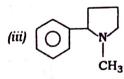


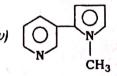


- (c) Which one of the following is not a terpene?
 - (i) Myrcene
 - (ii) Citral
 - (iii) Camphor
 - (iv) Quinine
- (d) The chemical structure of nicotine is









P23/993

(Continued)

- 2. Answer any four questions from the following: 2×4=8
 - (a) What is diazotization? How is benzene diazonium chloride prepared in the laboratory?
 - (b) Aromatic amines are weaker bases than aliphatic amines. Explain.
 - (c) Sulphonation of naphthalene gives α -isomer at low temperature and β -isomer at high temperature. Explain.
 - (d) Thiophene is more aromatic in nature than furan. Explain.
 - (e) What are the structural formulae of hygrine and conine?

Unit-I

3. Answer any three questions:

3×3=9

- (a) How would you distinguish among 1°, 2° and 3° amines with the help of nitrous acid?
- (b) Write short notes on any two of the following: $1\frac{1}{2}\times 2=3$
 - (i) Hofmann's exhaustive methylation
 - (ii) Hofmann elimination reaction
 - (iii) Gabriel phthalimide synthesis

P23/993

Discuss the synthesis of the following:

11/2×2=3

(i) Azobenzene from aniline

(ii) m-nitroaniline from nitrobenzene

Complete following reactions the 1×3=3 (any three):

(i)
$$\frac{1) \operatorname{Zn/NH_4Cl}}{2) \operatorname{Na_2Cr_2O_7/H_2SO_4}}$$
NO₂

(iii)
$$\bigcirc$$
 NH₂ + OHC \bigcirc \triangle

(iv)
$$\langle O \rangle$$
 $N_2^+CI^ Cu/HBr$

UNIT-II

4. Answer any three questions:

3×3=9

Complete the following reactions (any three): $1 \times 3 = 3$

(i) O
$$\frac{1) O_3}{2) H_2O/Zn}$$

(iii)
$$\leftarrow$$
 + H₂ $\xrightarrow{\text{CuO/Cr}_2O_3}$

(iv) OH + NH₃
$$\frac{(NH_4)_2SO_3}{150 \text{ °C/pressure}}$$

- Explain why, electrophilic substitution of anthracene and phenanthrene mainly takes place at 9 and 10 positions.
- How will you convert any two of the following? 11/4×2=3
 - (i) Naphthalene into α-naphthol
 - (ii) Phenanthrene into diphenic acid
 - (iii) Benzene into anthraquinone

P23/993

(d) Prepare naphthalene with the help of Haworth's synthesis. What happens when naphthalene is treated with hydrogen in presence of nickel catalyst?

2+1=3

UNIT-III

5. (a) Pyrrole, furan and thiophene are more reactive than benzene to electrophilic attack. Explain.

Starting with pyrrole, how will you get the following? 1+1=2

- (i) 2-pyrrole sulphonic acid
- (ii) Pyridine
- (c) Answer any three questions: 2×3=6
 - (i) Prepare pyridine by Hantzsch synthesis.
 - (ii) Write a short note on Knorr quinoline synthesis.
 - (iii) Complete the following reactions:

1+1=2

(2)
$$CH_2OH$$
 $H_2SO_4/FeSO_4$
 CH_2OH
 CH_2OH
 CH_2OH
 CH_2OH

P23/993

(Continued)

(iv) Convert the following:

1+1=2

- (1) Thiophene from sodium succinate
- (2) Furan from furfural
- (d) Explain why pyridine does not undergo Friedel-Crafts reactions. What happens when pyridine is treated with KNO₃ in presence of H₂SO₄ at 300 °C? 2+1=3

UNIT-IV

6. (a) Explain Zeisel method and Herzig-Meyer method with reference to the structure elucidation of alkaloids.

Or

Give one method of synthesis of hygrine.

- (b) Write down the sequential steps when Hofmann exhaustive methylation is employed with nicotine and name the product.
- (c) Give one medicinal use each of morphine and quinine. \(\frac{1}{2} \times 2 = 1 \)

Unit-V

7. (a) What is special isoprene rule? Explain with example.

P23/993

How will you show that citral is an α , β (b) unsaturated aldehyde?

Or

Draw the geometrical structure of citral.

- How will you synthesize the following (c) (any one)?
 - (i) Citral from methylheptenone
 - (ii) Geranic acid from geraniol